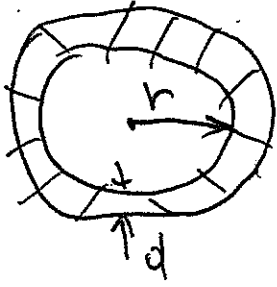


(Q4)



$$V = \frac{4}{3} \pi r^3 \rightarrow r = 28.8 \text{ cm}$$

for  $100 \text{ L} = V$

Assumptions : (i)  $d = 1 \text{ cm} \ll r = 28.8 \text{ cm}$ ,  
so curvature can be neglected.

(ii) There's equilibrium at the g-s interface

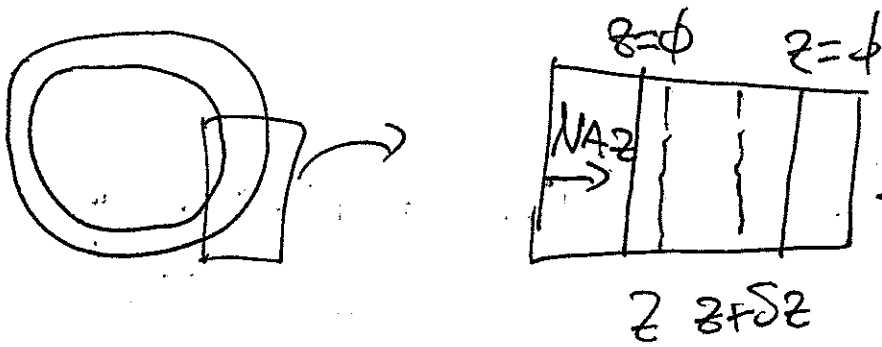
(iii) Ideal gas :  $P = cRT$  (1)  $\rightarrow$

$$\rightarrow \frac{dP}{dt} = RT \frac{dc}{dt} \text{ (2)}$$

(iv) Considering that  $C_{A0}$  is the initial  $\text{H}_2$  conc. in the tank,  $C_A$  is the conc. that has diffused through the walls, then:

$$C = C_{A0} - C_A \rightarrow \frac{dc}{dt} = \frac{dC_{A0}}{dt} - \frac{dC_A}{dt} \rightarrow$$

$$\rightarrow \frac{dc}{dt} = -\frac{dC_A}{dt} = -\frac{1}{V} \phi \quad \left( \begin{array}{l} \text{diffusion rate} \\ \text{of } \text{H}_2 \text{ through} \\ \text{the walls} \end{array} \right) \text{ (3)}$$



② Mass balance for the

$$S N_{A2}|_{z+\delta z} - S N_{A2}|_z = \phi \quad \begin{array}{l} \text{divide by} \\ S \delta z \\ \text{and } \delta z \rightarrow \phi \end{array}$$

$$\rightarrow \frac{dN_{A2}}{dz} = \phi \quad (4)$$

assumptions: steady state, no rxn,  
stainless steel is stagnant.

$X_A \ll 1$  due to diffusion in solid

Fick's law  $T, P$  constant inside the wall

$$N_{A2} = -c D_{AB} \frac{dX_A}{dz} + X_A (N_{A2} + N_{B2})$$

$$\rightarrow N_{A2} = -D_{AB} \frac{dC_A}{dz} \quad (5)$$

$$(4) \xrightarrow{(5)} \frac{d^2 C_A}{dz^2} = \phi \quad (6)$$

BCs

$$\text{for } z=0 \rightarrow C_A = C_{A0}$$

$$\text{for } z=d \rightarrow C_A = C_{A1} = \phi \text{ because it's vacuum outside the tank.}$$

Integrate (6) twice:  $\frac{d^2 C_A}{dz^2} = \phi \rightarrow$

$$\rightarrow \frac{dC_A}{dz} = C_1 \rightarrow C_A = C_1 z + C_2 \quad (7)$$

$$\text{from BC 1: } C_2 = C_{A0} \quad (8)$$

$$\text{from BC 2: } C_1 = \frac{C_{A1} - C_{A0}}{d} = \frac{-C_{A0}}{d} \quad (9)$$

$$(7) \xrightarrow[(9)]{(8)} C_A - C_{A0} = -C_{A0} \frac{z}{d} \quad (10)$$

$$\text{and (5)} \rightarrow N_{Az} = -D_{AB} \frac{dC_A}{dz} \xrightarrow{(10)}$$

$$\Rightarrow \boxed{N_{Az} = D_{AB} C_{A0} / d} \quad (11)$$

(b) The rate (in  $\frac{\text{gmol}}{\text{s}}$ ),  $\text{H}_2$  diffuses is equal to:  $N_{Az} * 4\pi r^2 \quad (12)$

$$\text{from (3) and (12)} \rightarrow \frac{dc}{dt} = -\frac{1}{\frac{4}{3}\pi r^3} \frac{D_{AB} C_{A0}}{d} * 4\pi r^2$$

$\rightarrow$

$$\rightarrow \frac{dc}{dt} = - \frac{3 P_{AB} C_{A0}}{r.d} \quad (13)$$

finally, (2)  $\xrightarrow{(13)}$

$$\boxed{\frac{dP}{dt} = - \frac{3 P_{AB} C_{A0} RT}{r.d}}$$