

# **Chemical additives in standardized food**

**Legislation**

**Food & Beverages  
Code**

**For all additives, please know  
the following:**

- **Definition**
- **1-2 Examples  
of Compounds**
- **Mode of Action**

# Chemical additives



## Chemical additives:



**«Substances added to food for the purpose of preserving flavor, taste, appearance, and shelf life»**

- synthetic or of natural origin
- For example, vinegar & salt: their use has been known for centuries

## Numbering:

- For the convenience of legislation and public information: international number E (*Codex Alimentarius, 1963, FDA & FAO*)



- Outside Europe, the prefix E is usually omitted.
- In America, substances approved for use in food are characterized as 'Generally recognized as safe' or "**GRAS**" (FDA)



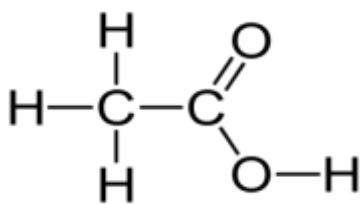
# The main categories of chemical additives in standardized food



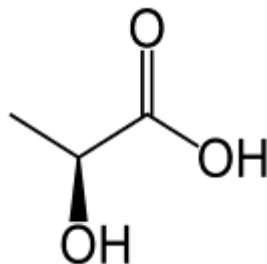
# 1. Acids - Acidulants

«Substances that increase the acidity of food and/or impart sour taste»

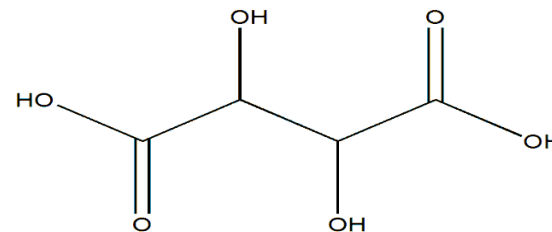
- They provide a '**sour**' and '**intense**' flavor
- They function as both **preservatives** and **antioxidants**



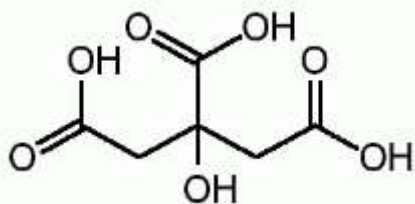
$\text{CH}_3\text{COOH}$   
**Oxalic acid**



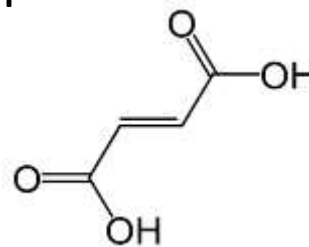
$\text{CH}_3\text{-CH(OH)-COOH}$   
**Lactic acid**



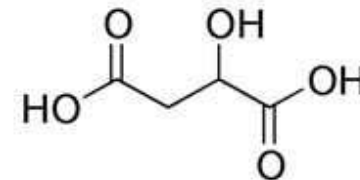
$\text{HOOC-CH(OH)-CH(OH)-COOH}$   
**Acetic acid.**



$\text{HOOC-CH}_2\text{-C(COOH)(OH)-CH}_2\text{-COOH}$   
**Citric acid**



$\text{HOOC-CH=CH-COOH}$   
**Fumaric acid**



$\text{HOOC-CH}_2\text{-CH(OH)-COOH}$   
**Malic acid**

## 2. Acidity regulators

*«Substances that alter or control the acidity or alkalinity of food»*

- **organic or inorganic acids**
- **alkalis**
- **compounds with buffering capacity**

**For example, citric acid, oxalic acid, and lactic acid**

### 3. Anticaking agents

*«Substances that reduce the tendency of individual food particles to adhere to each other»*

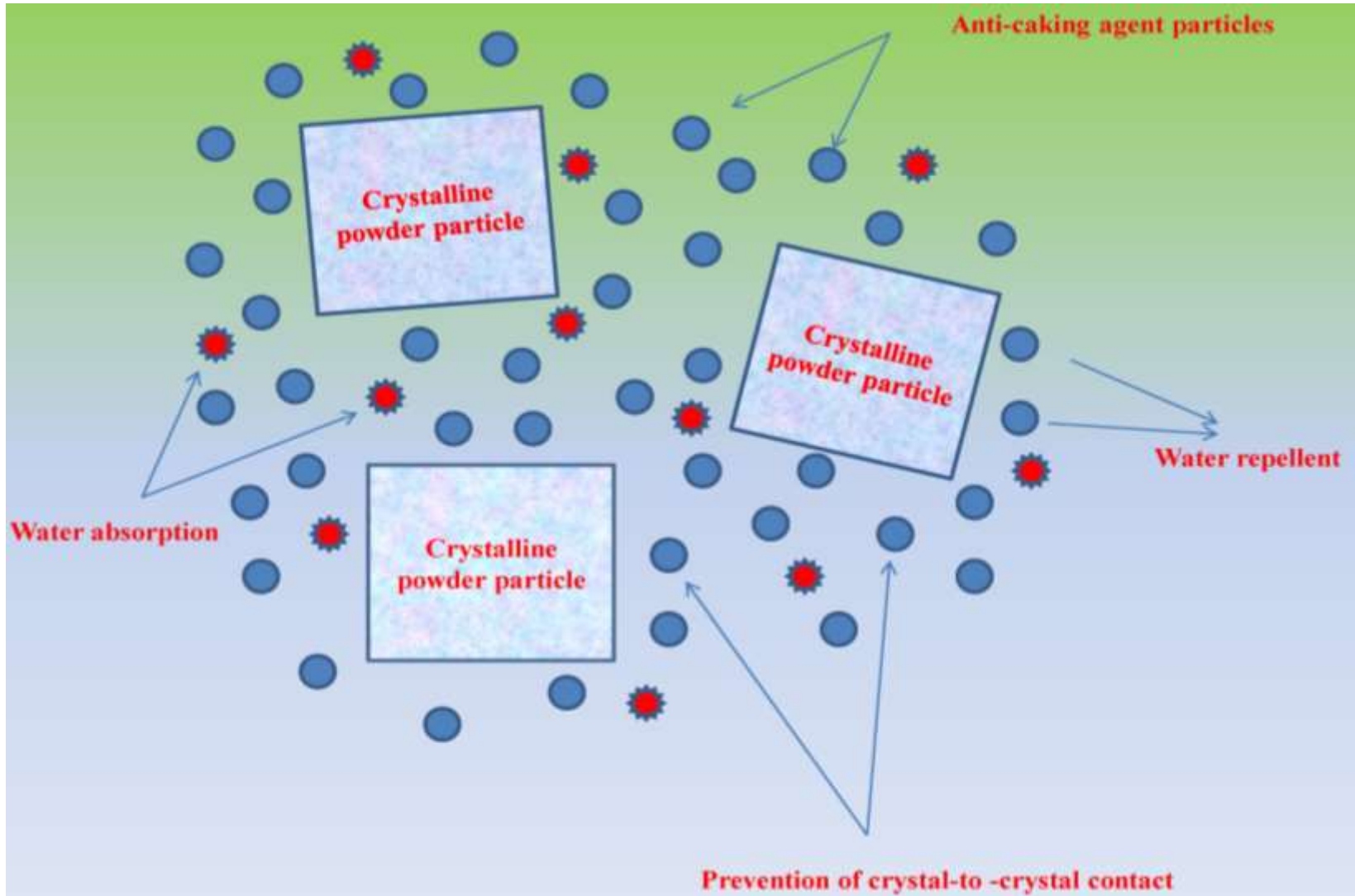
- They are commonly used in powdered or granulated food products such as milk powder, tea, coffee, sugar, salt, etc.

**Examples include:**

- ✓ Ca & Mg phosphates
- ✓ Na, K, Ca, Mg, Al, Zn pyrophosphates
- ✓ Talc, kaolin, bentonite
- ✓ Na, Ca, K, Mg salts of fatty acids (which are also used as emulsifiers)
- ✓ Magnesium oxide
- ✓ K, Ca, Na ferrocyanides



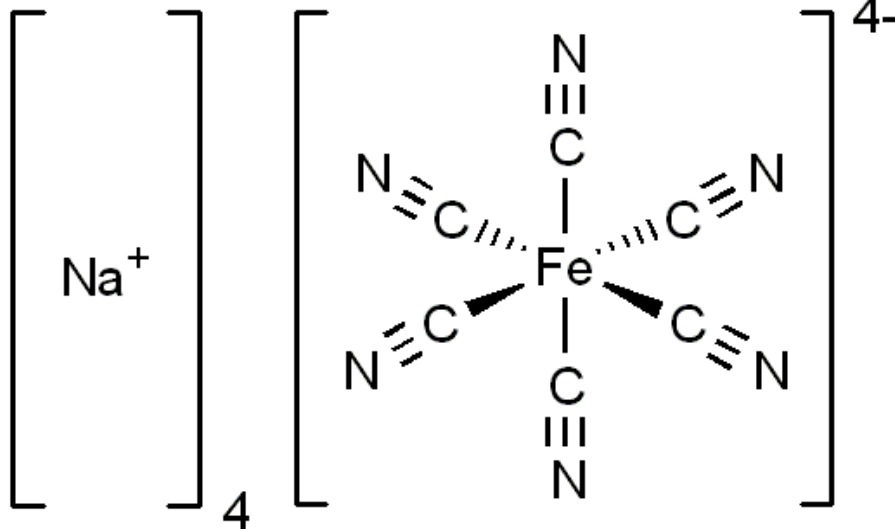
### 3. Anticaking agents



(mode of action)

### 3. Anticaking agents

For example **sodium ferrocyanide**



- Chemical formula:  $\text{Na}_4\text{Fe}(\text{CN})_6$  &  $\text{Na}_4\text{Fe}(\text{CN})_6 \cdot 10\text{H}_2\text{O}$  (Na hexacyanoferrate or Yellow Prussiate of Soda-YPS)
- Non-toxic compound
- But in under acidic conditions or in the presence of light, it can decompose and release hydrogen cyanide (HCN)

### 3. Anticaking agents

## Sodium ferrocyanide in salt (NaCl)

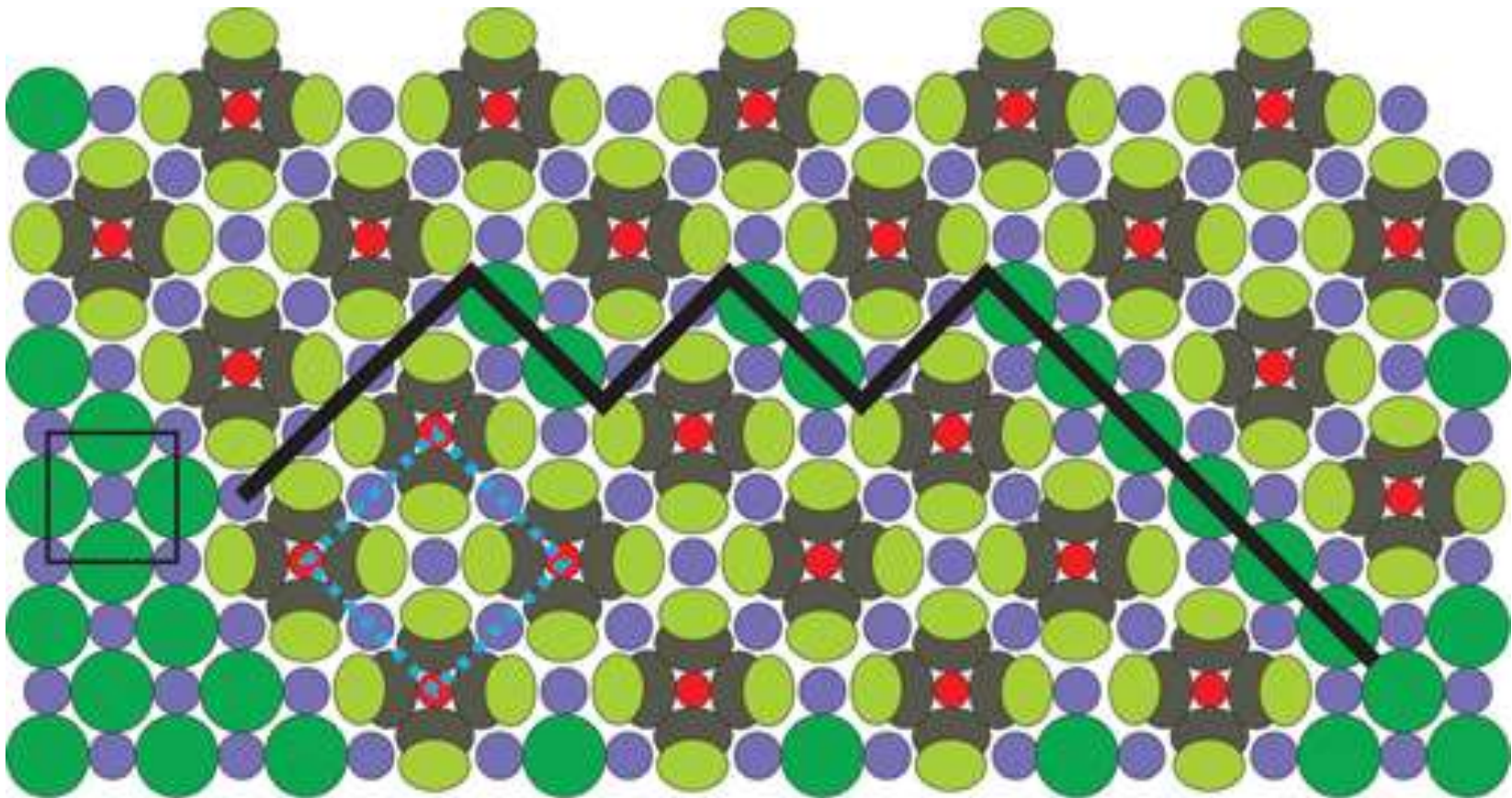
**Possible Mechanism** (*Bode et al., Cryst. Growth Des. 2012, 12, 4, 1919–1924:*

- NaCl crystals have a strong tendency to agglomerate.
- Through XRD analysis, it was found that a  $\text{Fe}(\text{CN})_6^{-4}$  ion is adsorbed on the surface of the NaCl crystal, replacing one  $\text{Na}^+$  ion and five surrounding  $\text{Cl}^-$  ions.
- The coverage is approximately 50%.
- Due to the charge of the adsorbed  $\text{Fe}(\text{CN})_6^{-4}$  ions on the surface of NaCl, the crystal can only continue to grow from an energetically unfavorable  $\text{Na}^+$  position or by releasing a  $\text{Fe}(\text{CN})_6^{-4}$  ion.
- Therefore,  $\text{Fe}(\text{CN})_6^{-4}$  effectively inhibits further growth of NaCl crystals, preventing them from forming agglomerates.

### 3. Anticaking agents

## Sodium ferrocyanide in salt (NaCl)

**Possible Mechanism** (*Bode et al., Cryst. Growth Des. 2012, 12, 4, 1919–1924:*

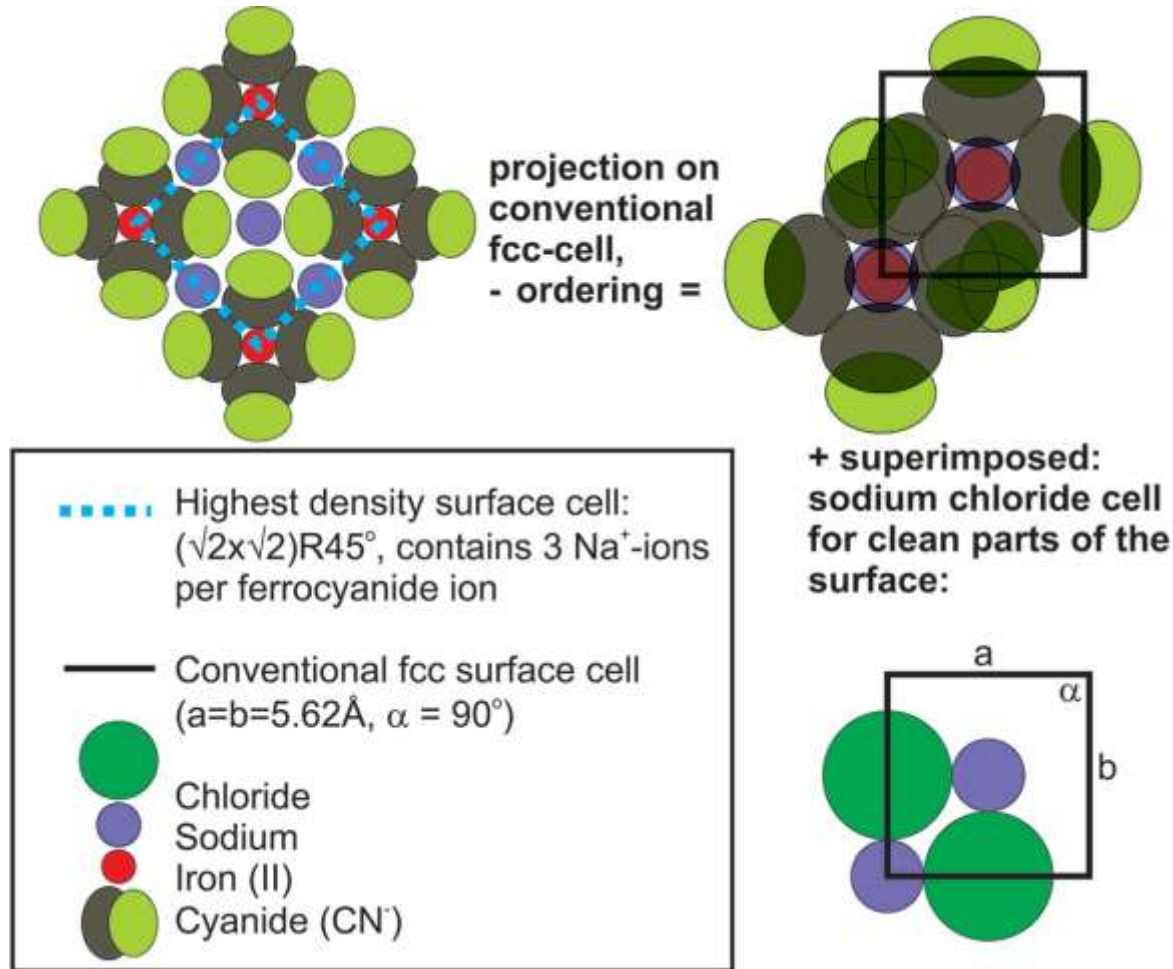




### 3. Anticaking agents

## Sodium ferrocyanide in salt (NaCl)

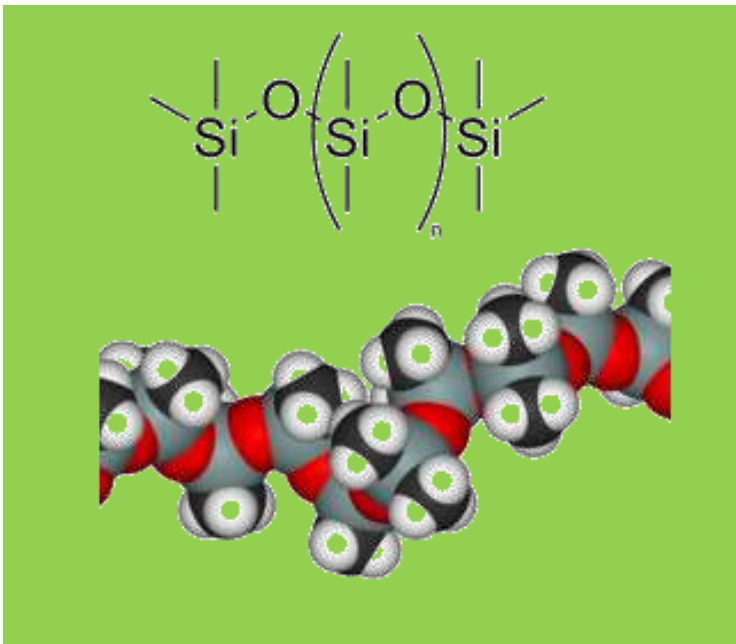
**Possible Mechanism** (*Bode et al., Cryst. Growth Des. 2012, 12, 4, 1919–1924:*



## 4. Antifoaming agents

*“Substances that prevent or limit foam formation”*

- Applications: Soft drinks, Frying oils, etc.

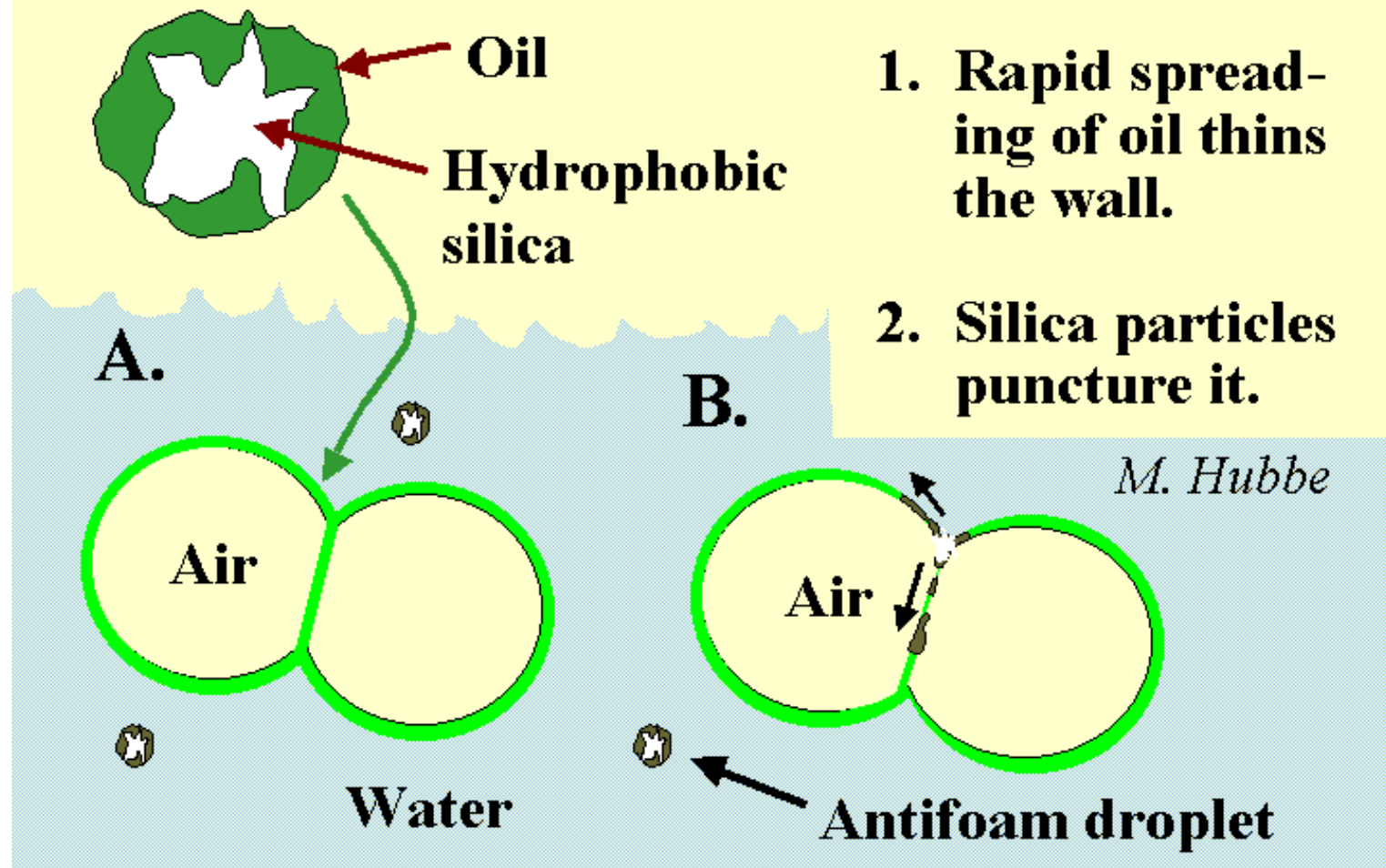


**Dimethyl polysiloxane -  
silicone oil**

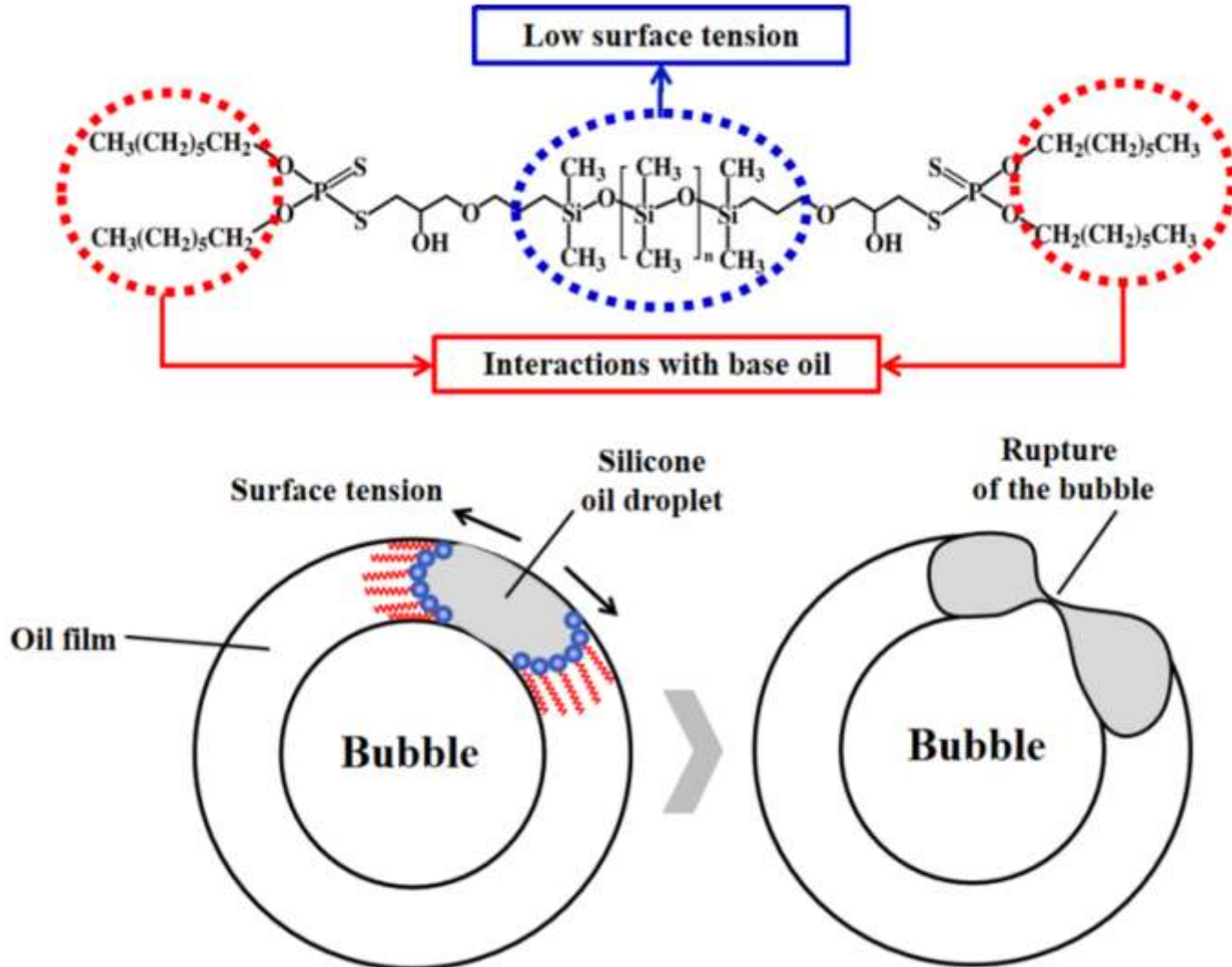


## 4. Antifoaming agents

### Anti-Foam Schematic



## 4. Antifoaming agents



Antifoaming mechanism of chemically modified silicone oil (**diocetyl dithiophosphate-terminated silicone oil**)  
 (Luan et al., 2022; <https://doi.org/10.3390/lubricants10120364>)

## 5. Bulking agents

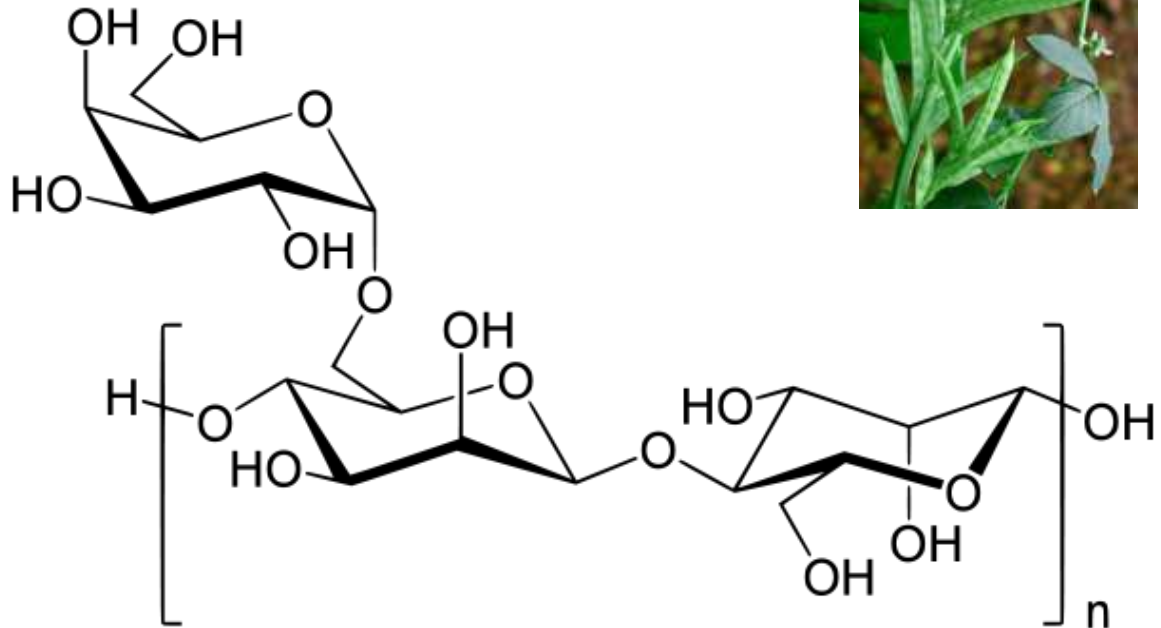
*«Substances that contribute to the increase of the volume or weight of food without significantly adding to its caloric value»*

- **Applications: meat substitutes, low-calorie foods, confectionery and bakery products, and most standardized foods**
  - ✓ Guar gum
  - ✓ Carnauba wax
  - ✓ Mannitol
  - ✓ Maltitol
  - ✓ Glycerol
  - ✓ Methyl cellulose
  - ✓ Polydextrose
  - ✓ Pectin
  - ✓ etc.

## 5. Bulking agents

### guar gum (Galactomannan):

linear chain of  $\beta$ -1,4-linked mannose monomers with  $\alpha$  1,6-galactose branches every 2 mannose units

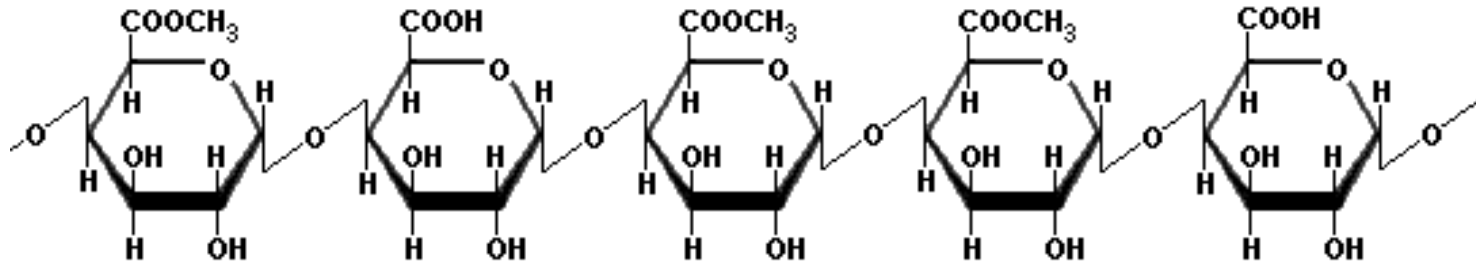


**Guaran unit**

[https://en.wikipedia.org/wiki/Guar\\_gum](https://en.wikipedia.org/wiki/Guar_gum)

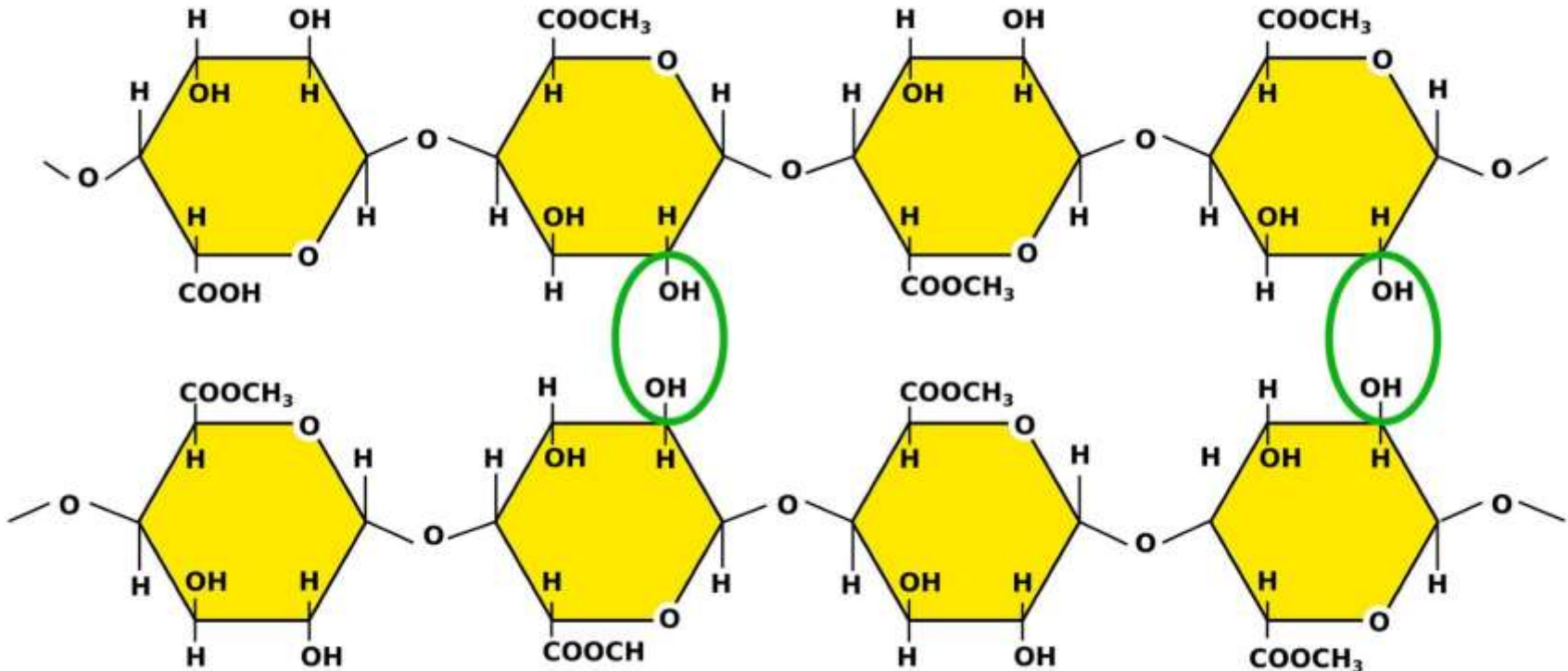
## 5. Bulking agents

**Pectin** - polymer of  $\alpha$ -galacturonic acid & its methyl esters (partially esterified with methanol)



## 5. Bulking agents

### Pectin gel formation mechanism



**Pectin with a high degree of esterification (high-methoxyl pectin; HM pectin)  
→ Forms intra-molecular H-bonds**

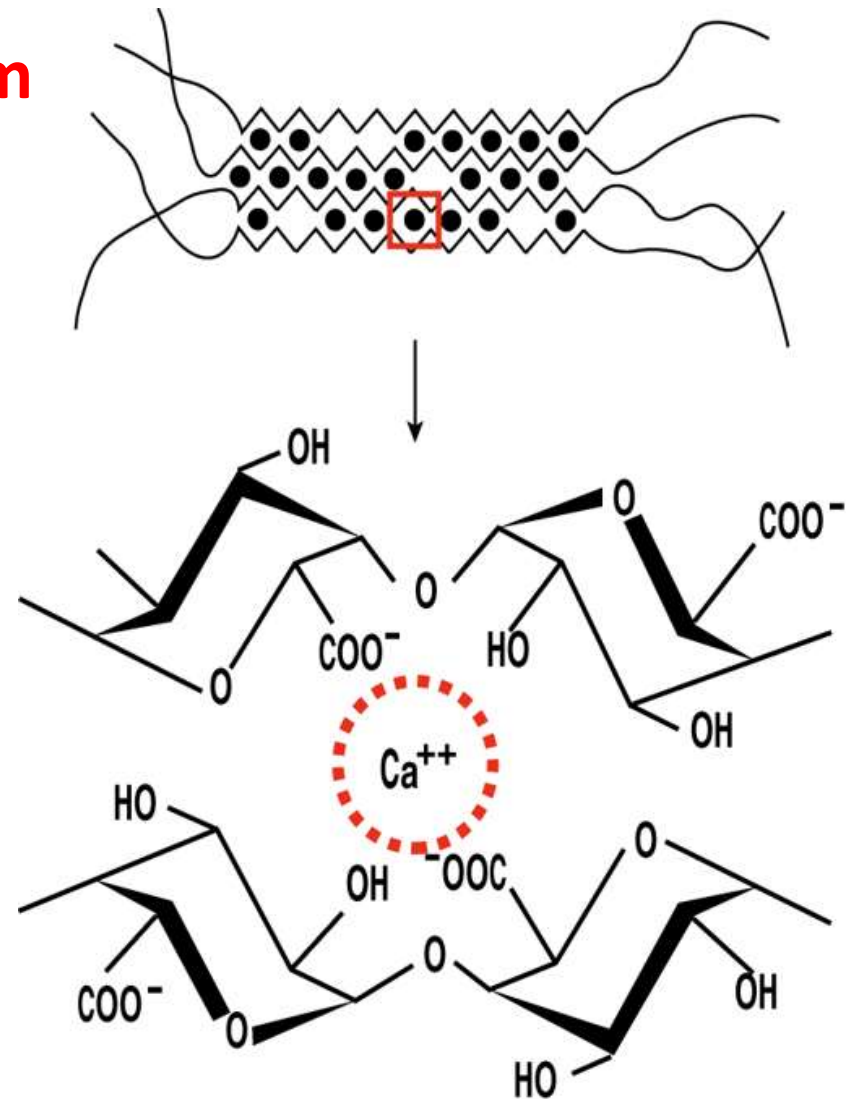


## 5. Bulking agents

### Pectin gel formation mechanism

**Pectin with a low degree of esterification (low-methoxyl pectin; LM pectin)**

→ Forms intra-molecular H-bonds, as well as cross-linking with divalent cations, mainly  $\text{Ca}^{2+}$ , following the "Egg Box" model:



## 6. Emulsifiers

*“Substances that allow the formation or maintenance of a homogeneous mixture of two or more immiscible phases, such as oil in water”*

- ✓ Salts of fatty acids with Al, Ca, Na, Mg, K & NH<sub>4</sub>
- ✓ Diphosphate, triphosphate, and polyphosphate salts
- ✓ **Sorbitans**: Monostearate, Tristearate, Monooleate, Trioleate, Monopalmitate, Monolaurate
- ✓ **Polysorbates**: Monostearate (p.60), Tristearate (p.65), Monooleate (p.80), Monopalmitate (p.40), Monolaurate (p.20)

## 6. Emulsifiers

- ✓ Oxalic, lactic, citric, and tartaric esters of mono- and diglycerides of fatty acids
- ✓ Esters of fatty acids with sucrose
- ✓ Saccharoglycerides
- ✓ Beta-cyclodextrin
- ✓ Cellulose, methyl-, ethyl-, hydroxypropyl-, ethylhydroxyethyl-, carboxymethyl-cellulose, hemicelluloses
- ✓ Various gums
- ✓ Peptides
- ✓ 1,2-Propanediol and its esters with fatty acids

## 6. Emulsifiers

The oldest emulsifier is **Egg yolk**

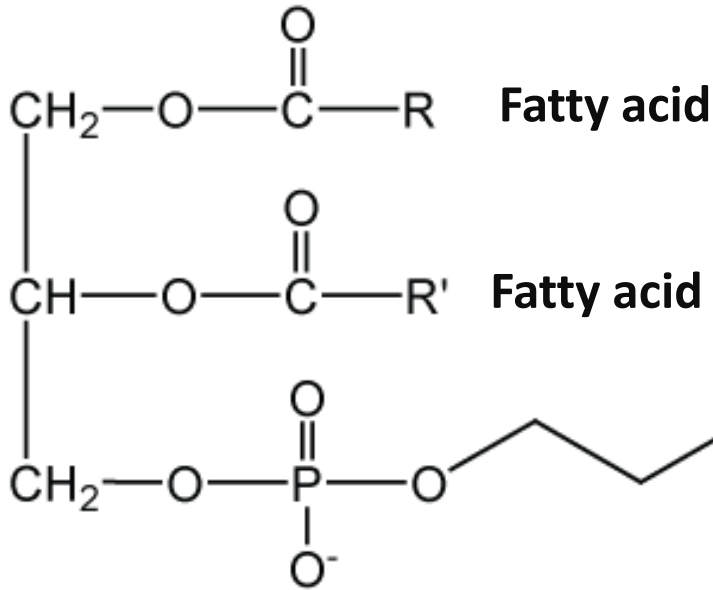
Its main emulsifying substance is lecithin, which is a mixture of fatty substances composed of:

- Phosphoric acid
- Choline
- Glycerides
- Phospholipids such as phosphatidylcholine, phosphatidylethanolamine, and phosphatidylinositol



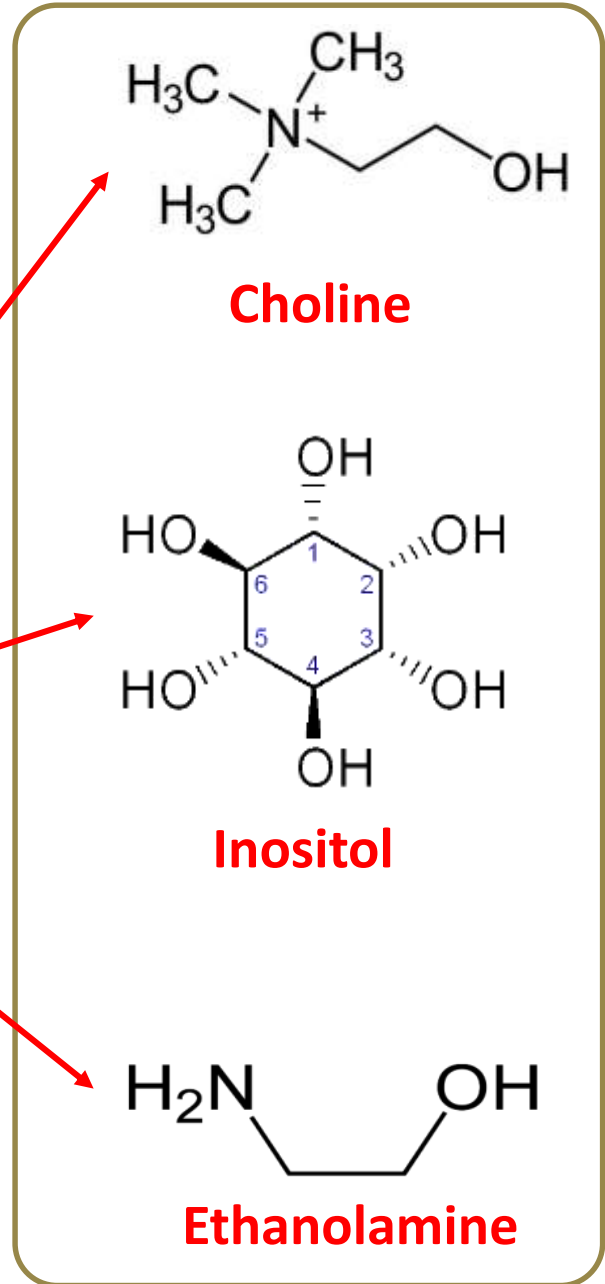
# 6. Emulsifiers

## lecithin



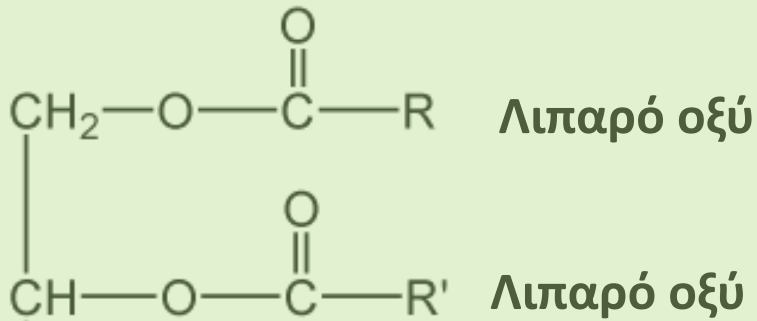
### Phospholipid

**X** = Choline, ethanolamine or inositol

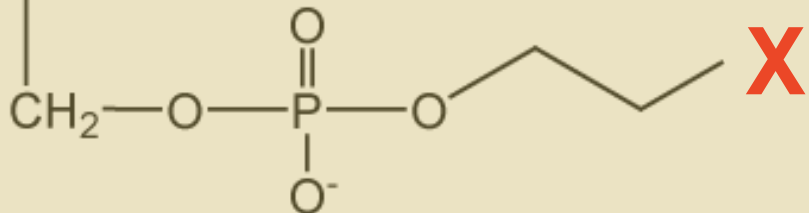


## 6. Emulsifiers

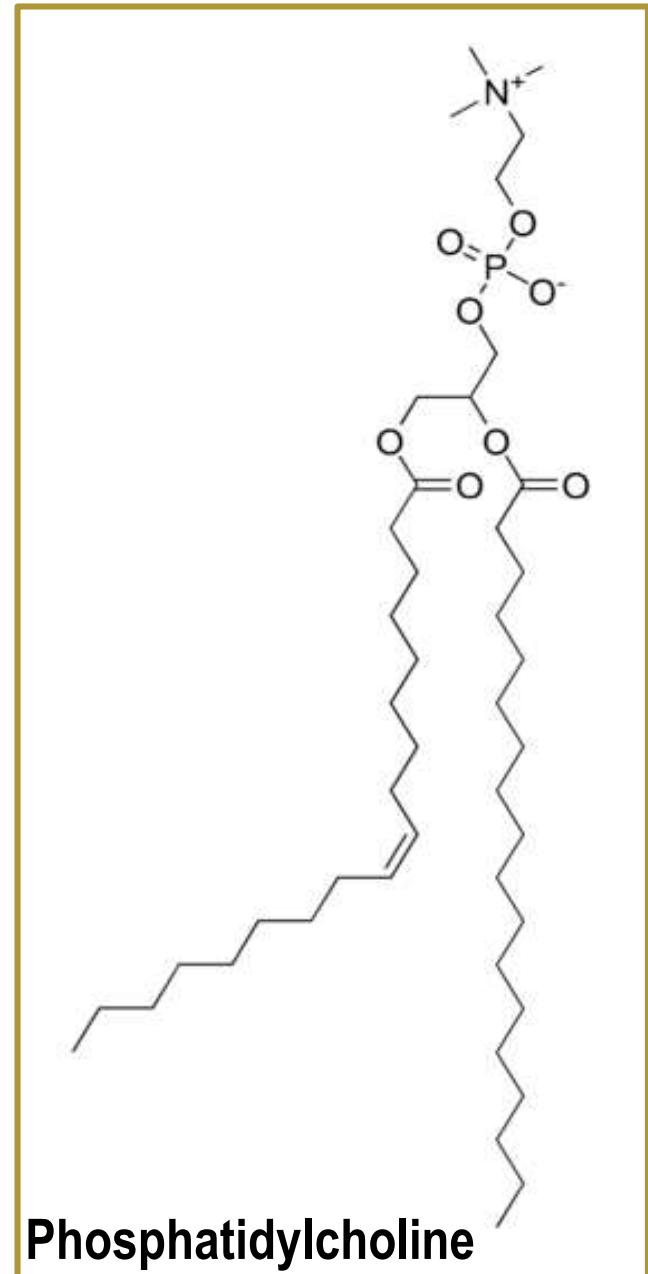
### lecithin



Hydrophobic region



Hydrophilic region

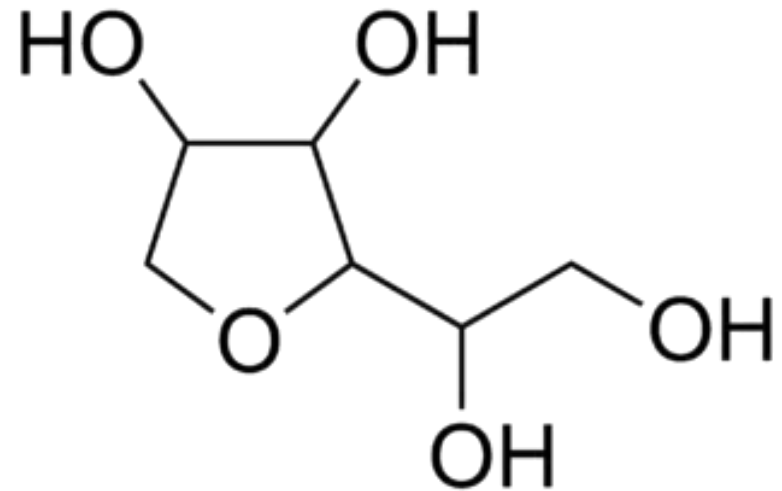




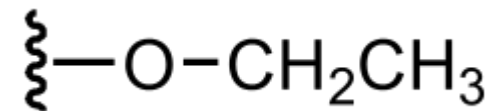
## 6. Emulsifiers

### Polysorbates

✓ Oily liquids derived from ethoxylated sorbitan (a derivative of sorbitol), esterified with fatty acids



**Sorbitan**

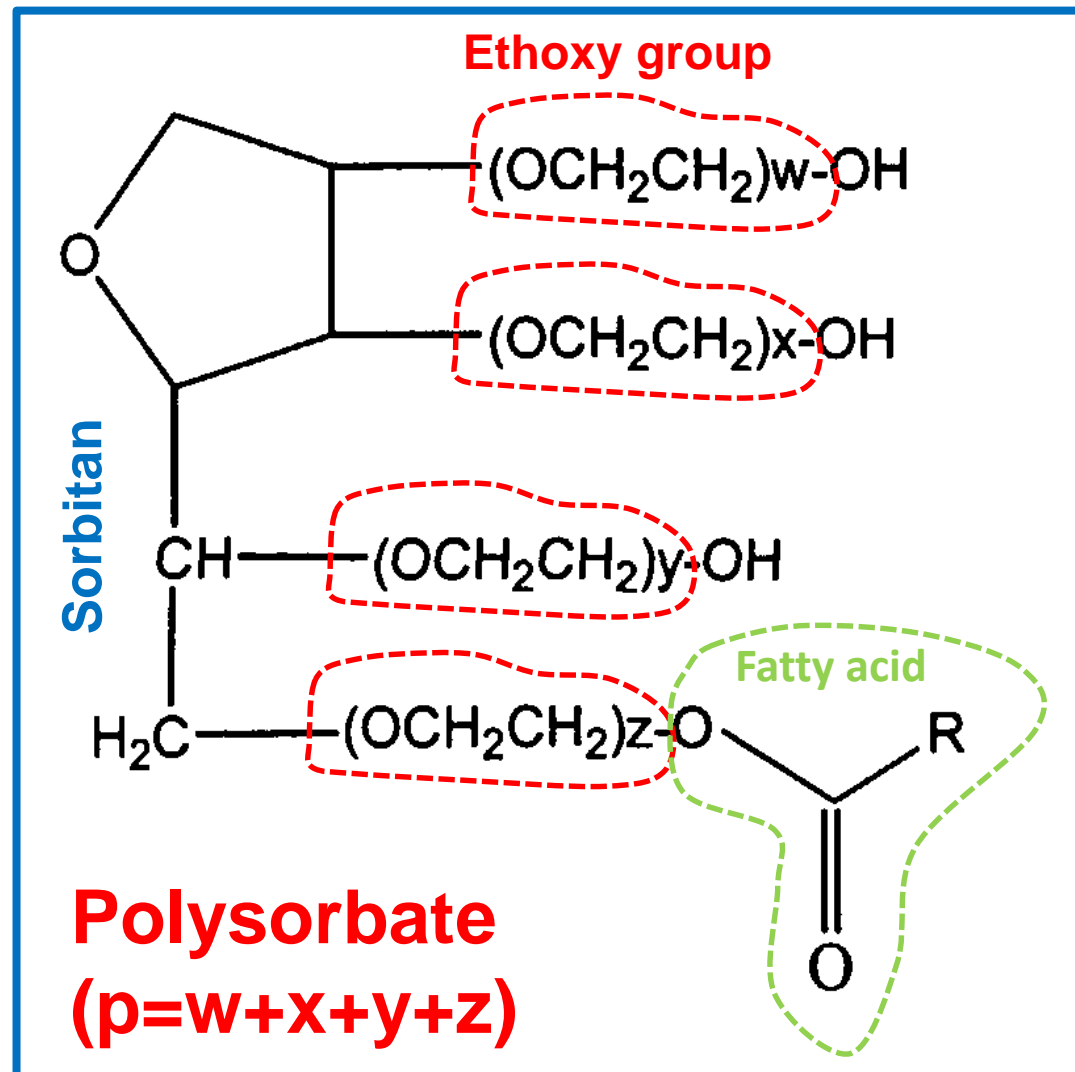


**Ethoxy group**

## 6. Emulsifiers

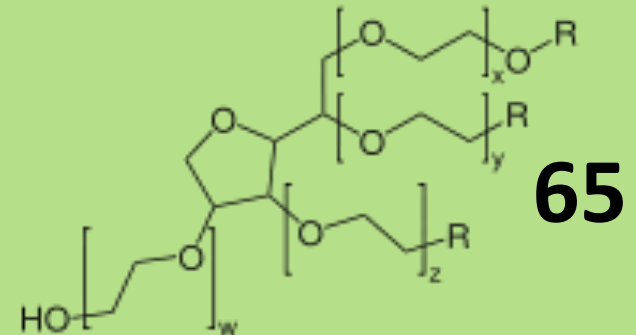
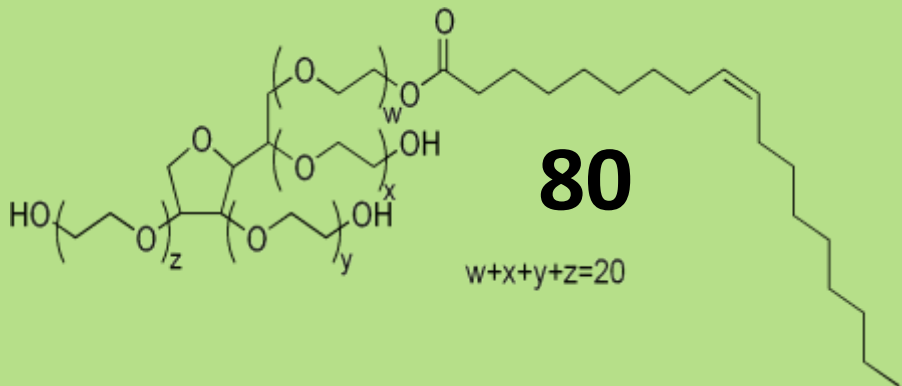
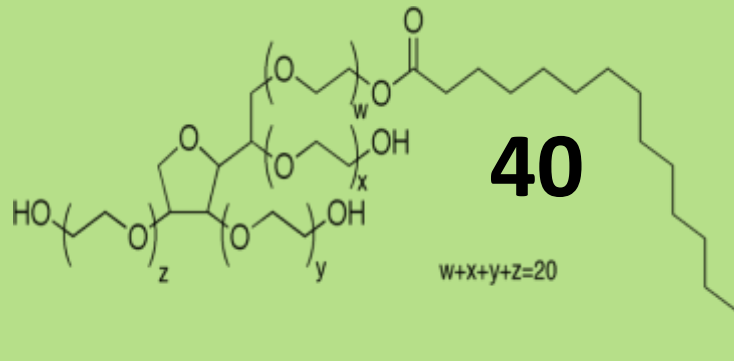
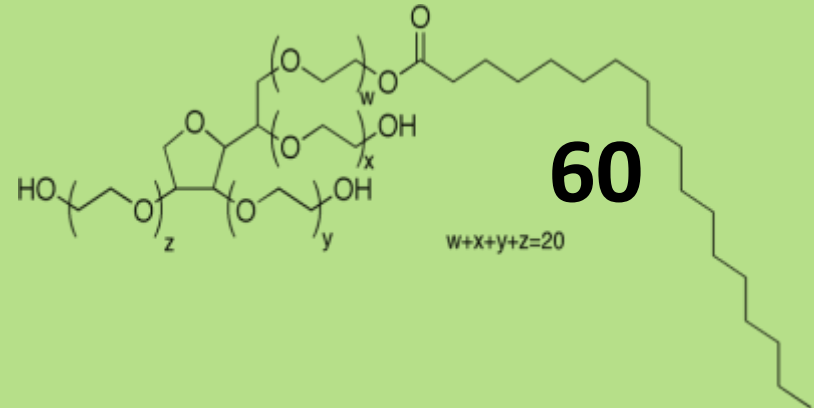
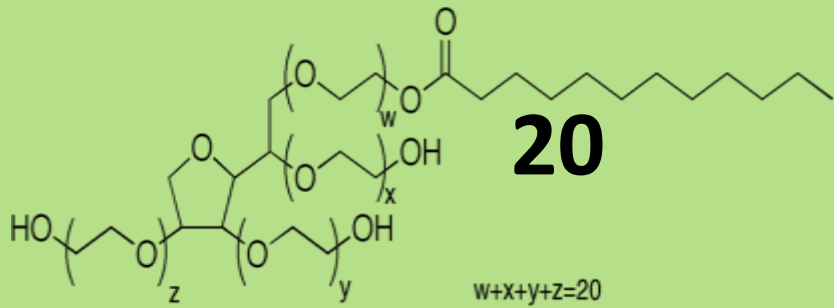
### Polysorbates

- ✓ Monostearate (p.60)
- ✓ Tristearate (p.65)
- ✓ Monooleate (p.80)
- ✓ Monopalmitate (p.40)
- ✓ Monolaurate (p.20)

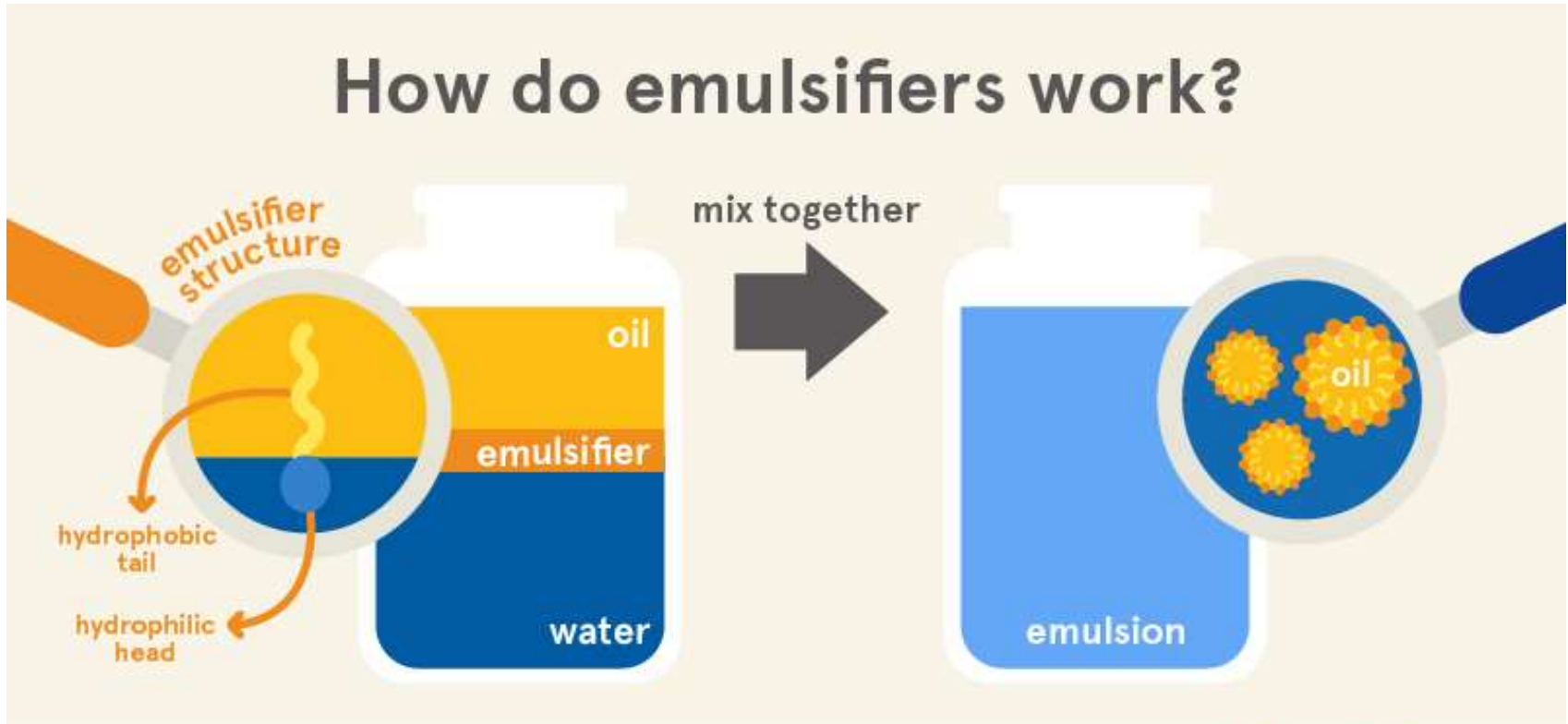


# 6. Emulsifiers

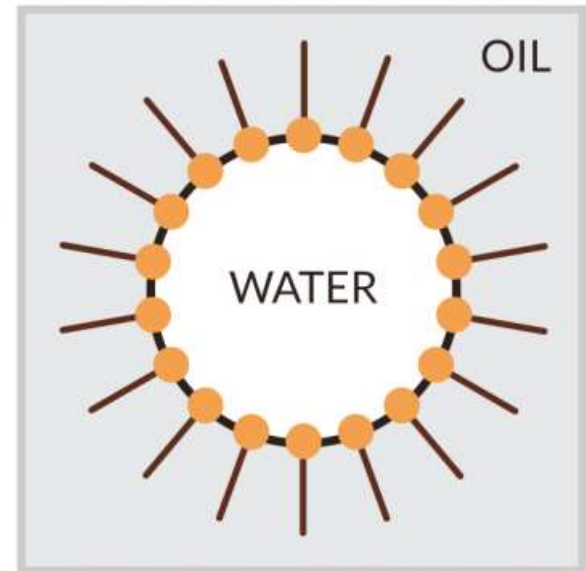
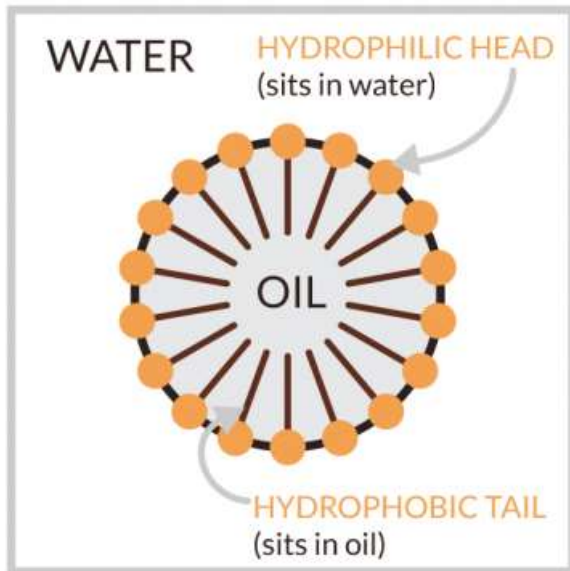
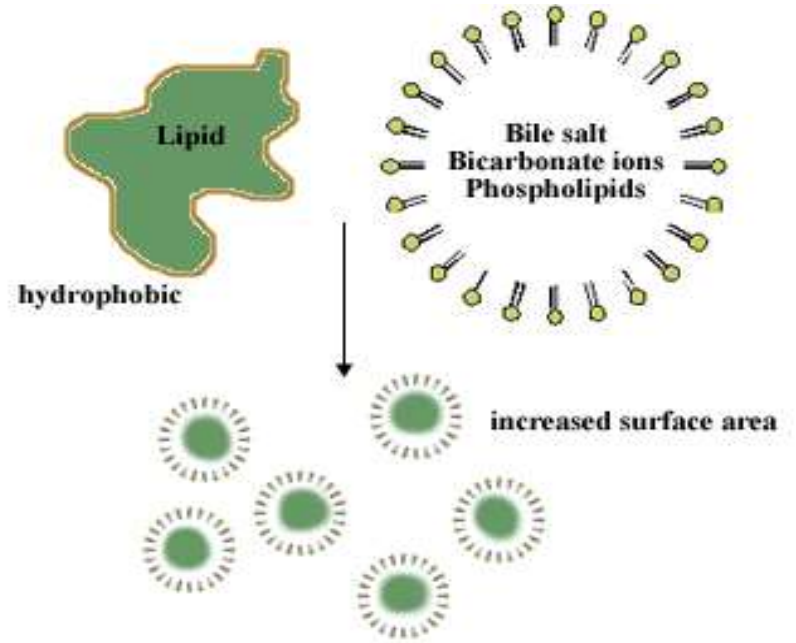
## Polysorbates



## 6. Emulsifiers



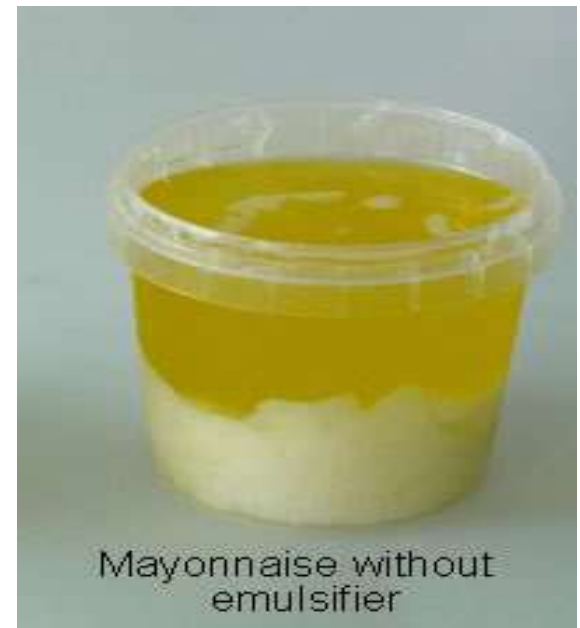
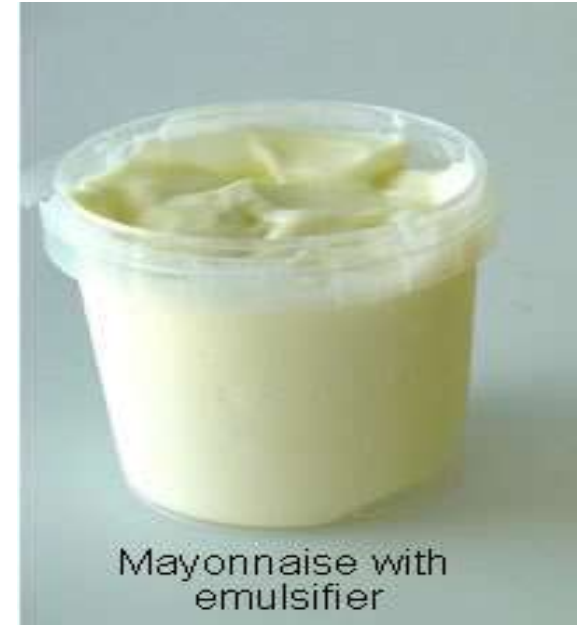
# 6. Emulsifiers



## 6. Emulsifiers

### application

- Margarine
- Mayonnaise
- Chocolate
- Ice cream
- Bread
- Desserts/Creams/Whipped cream/Mousse
- Biscuits/Cookies
- Soft drinks/Beverages
- Caramels
- Chewing gum
- Etc.





## 7. Hardeners

*“substances that render or maintain the tissues of fruits or vegetables firm or crisp, or interact with coagulating agents for the preparation or enhancement of coagulation”*

- **Calcium hydrogen sulphite (E227)**

Beer, canned fruits/vegetables, jams, fruit juices, etc.

- **Calcium citrate (E333)**

Ice cream, soft drinks, jams, condensed milk, cheeses.

- **Calcium phosphate (E341)**

Canned fruits, powdered foods.



## 7. Hardeners

- **Calcium sulfate (E516)**

Beer

- **Calcium chloride (E509)**

Beer, canned fruits and vegetables, pickles, cheeses

- **Chloride (E511) and Magnesium sulfate (E518)**

Cheeses

- **Aluminum sulfate (E520)**

Cheeses

- **Calcium hydroxide (E526)**

Cheeses

- **Calcium gluconate (E578) and Magnesium gluconate (E580)**

Canned vegetables



## 8. Gelling agents

*“Substances that give texture to a food by forming a gel”*

- **Polysaccharides**  
(Starch, gums)
- **Proteins**  
(Gelatin, egg albumin, collagen)
- **Flours**



## 9. Thickening agents

*“Substances that increase the viscosity of a food”*

- Alginate acid & salts with Na, Ca, K
- Agar
- Carrageenan
- Gums: oat, guar, acacia, tragacanth, xanthan, gelatin, etc.
- Gelatin
- Dextrin, Maltodextrin
- Starch & Modified starches
- Bakery dough glucan
- Arabinogalactan

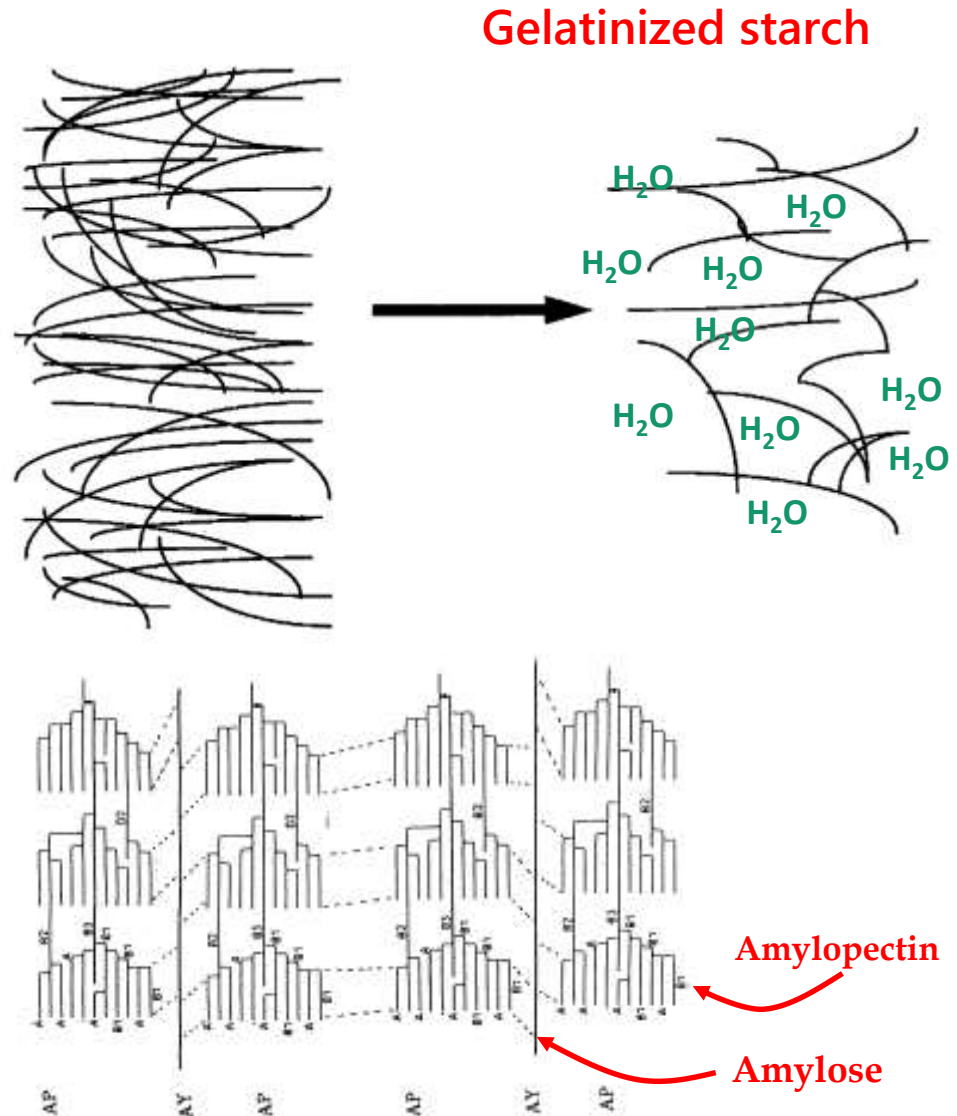


## 8,9. Gelling agents/thickening agents

### Starch gels

When starch grains are in the presence of water:

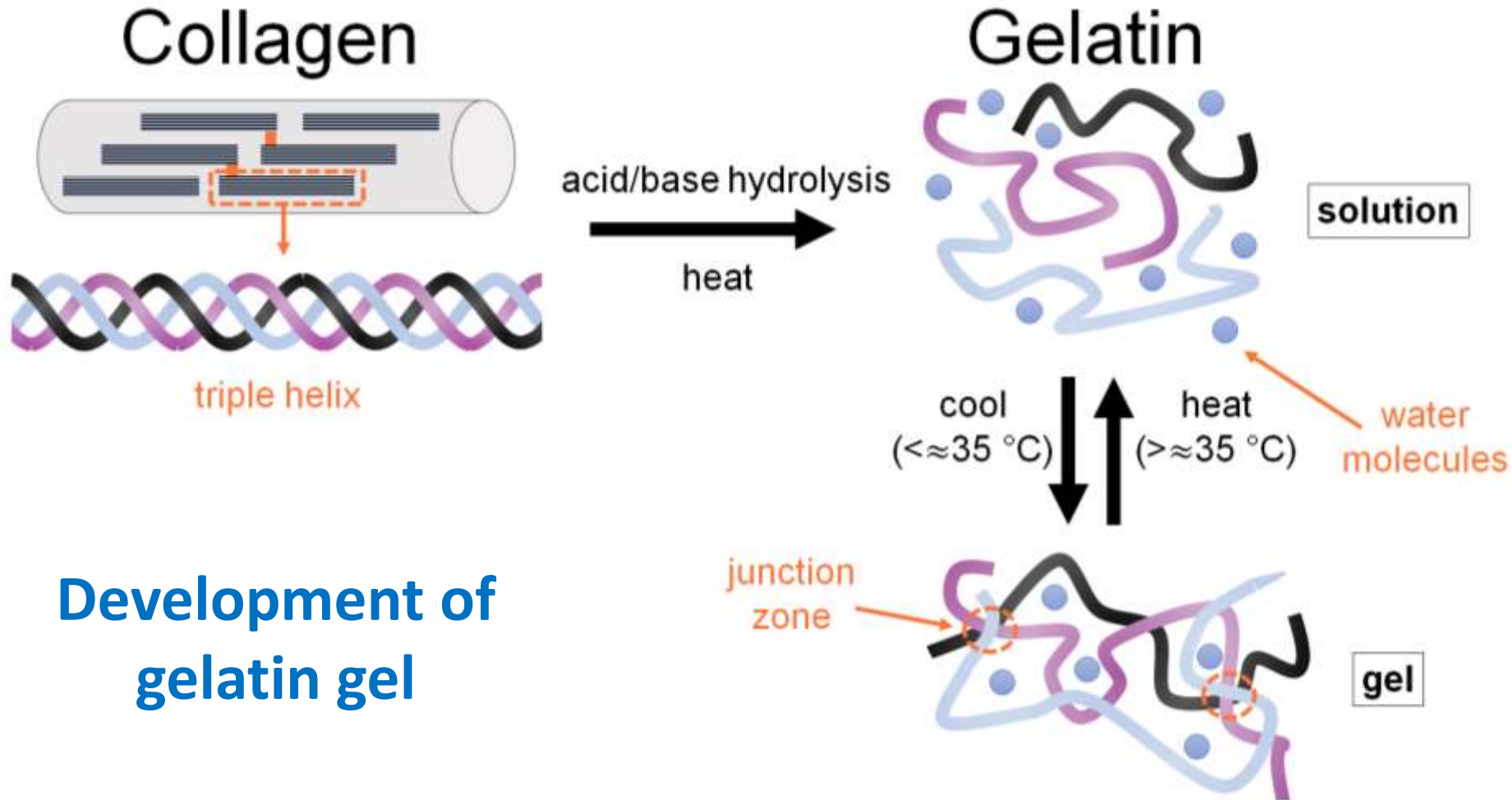
- They undergo slight swelling while maintaining their structure.
- With increasing temperature, **intermolecular H-bonds** are formed.
- The activity of water is significantly reduced, and the viscosity increases.
- The crystalline structure of starch is lost, and it undergoes **gelatinization**.





## 8,9. Coagulating agents/thickening agents

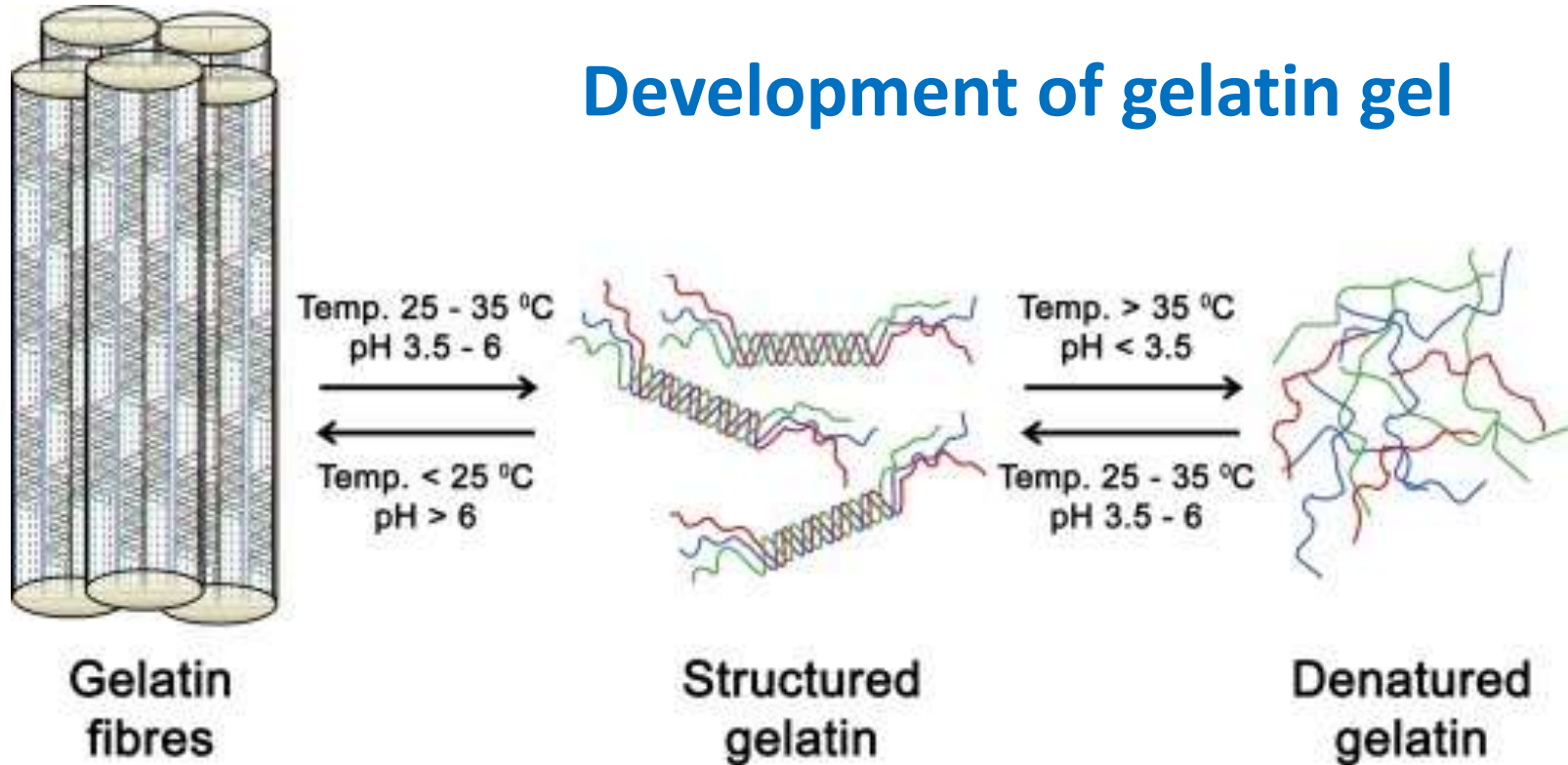
### Protein gels: Gelatin



Development of gelatin gel

## 8,9. Coagulating agents/thickening agents

### Protein gels: Gelatin



- The transformation of **collagen** produces thermally reversible **gelatin**.



## 10. Glazing agents

*“Substances applied to the external surface of food to provide a shiny appearance or a protective coating”*

**E901 - Beeswax**

**E902 - Candelilla wax**

**E903 - Carnauba wax**

**E904 - Shellac**

**E905 - Microcrystalline wax (Petroleum wax, Paraffin wax)**

**E912 - Montanic acid esters (Montan wax)**

**E914 - Oxidized polyethylene wax**



## 11. Humectants

*“Hygroscopic substances that prevent the drying of food by reacting to a low humidity environment or promoting the dissolution of a powder in a liquid medium”*

**E965 - Maltitol**

**E967 - Xylitol**

**E420 - Sorbitol**

**E422 - Glycerol**

**E1518 - Glyceryl triacetate**

**E1520 - Propylene glycol**

**E1200 - Polydextrose**

### **Applications:**

- Starch/jelly creams
- Toppings
- Sauces
- Salad dressings
- Caramels
- Bakery products
- Ice creams
- Etc.



## 12. Modified starches

*“Substances derived from edible starches that have undergone one or more chemical modifications. They may have undergone physical or enzymatic treatment and could have undergone acid or alkaline thinning or bleaching”*

### Applications as:

- **Thickening agents**
- **Gelling agents**
- **Stabilizers**
- **Emulsifiers**



## 12. Modified starches:

### **E1404 - Oxidized starch**

Starch treated with sodium hypochlorite

### **E1410 – Starch dihydrogen phosphate**

Starch esterified with orthophosphoric acid, Na or K orthophosphate, or Na tripolyphosphate

### **E1412 - Acid-treated starch**

Cross-linked with Na trimetaphosphate or phosphoryl chloride

### **E1413 - Phosphorylated distarch phosphate**

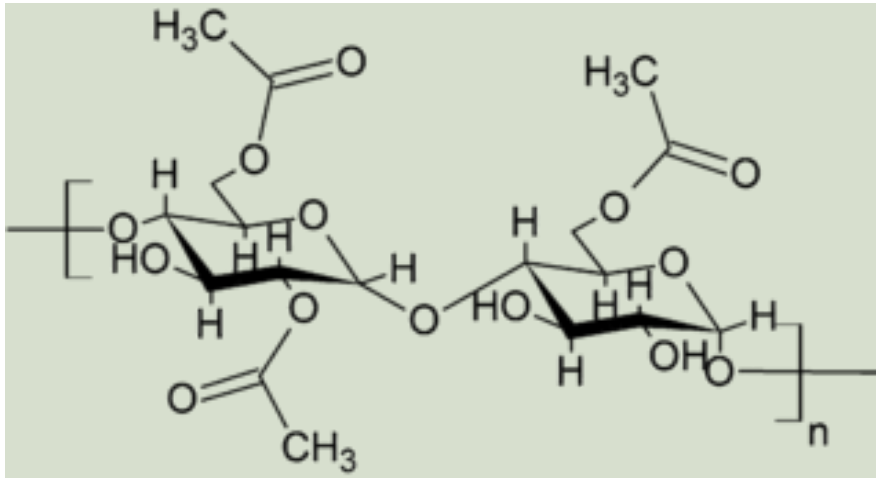
Combination of treatments such as in E1410 and E1412

### **E1414 - Acetylated distarch phosphate**

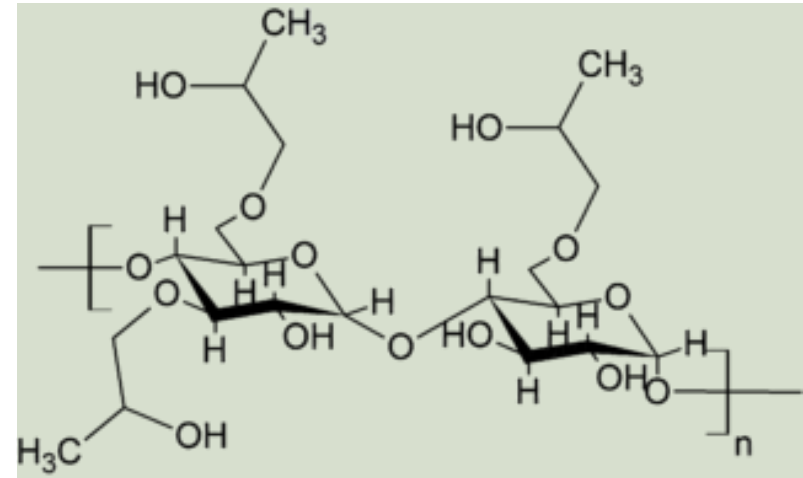
Cross-linked with Na trimetaphosphate or phosphoryl chloride and esterified with anhydrous citric acid or vinyl ester

And many more...

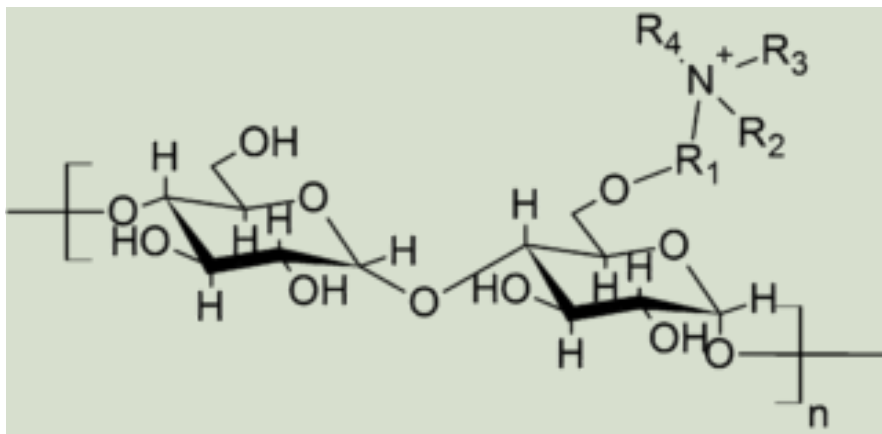
## 12. Modified starches



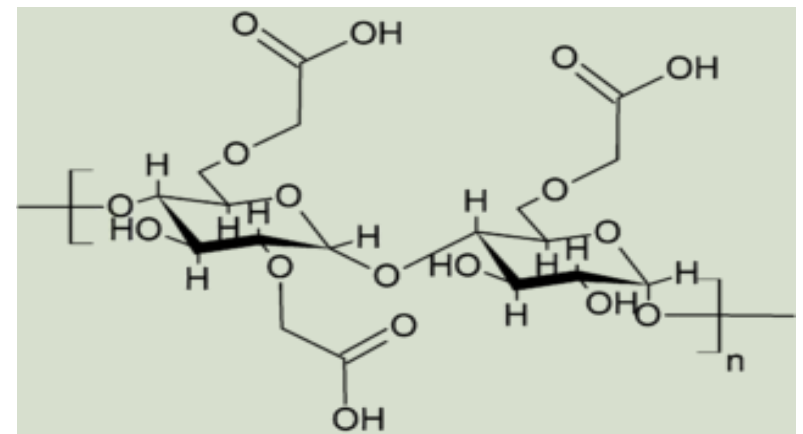
**Oxidized starch**



**Hydroxypropyl starch**



**Cationic starch**



**Carboxy Methyl Starch (CMS)**



## 13. Packaging gases

“gases other than air that are introduced into a container before, during, or after placing food in said container”  
(modified atmosphere environment)

E938 Argon

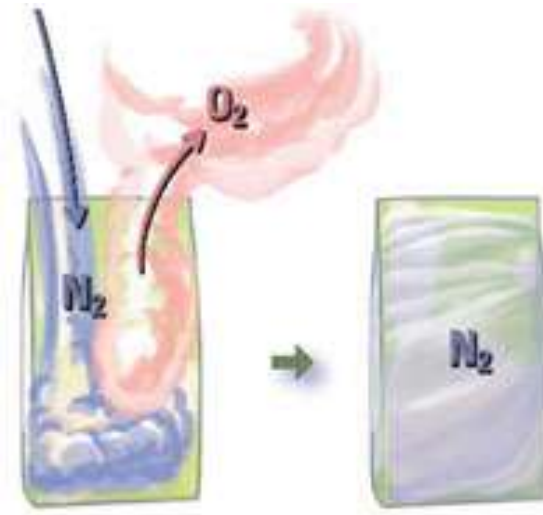
E939 Helium

E940 Dichlorodifluoromethane (\*)

E941 Nitrogen

E948 Oxygen

E949 Hydrogen (\*)



\*repealed either for being explosive or for depleting the ozone layer



## 14. Propellants

“Gases other than air that cause the expulsion of food from a container”

E941 Nitrogen

E942 Nitrous oxide

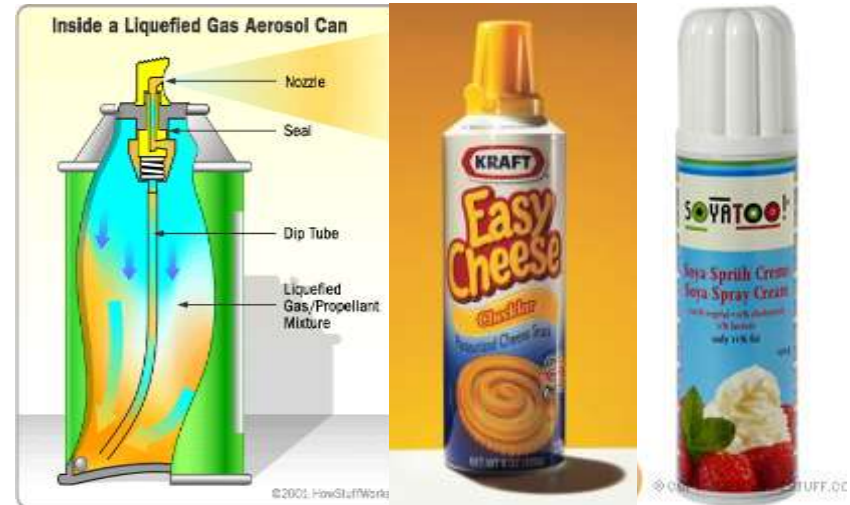
E943a Butane (\*)

E943b Isobutane (\*)

E944 Propane (\*)

E945 Chloropentafluoromethane (\*)

E946 Octafluoromethane (\*)



\*repealed either for being explosive or for depleting the ozone layer

## 15. Raising (or leavening) agents in baking

“Substances or combinations of substances that release gas and therefore increase the volume of the dough or batter”

A) Chemical substances (pharmaceutical grade) of an alkaline nature that "contain" (release)  $\text{CO}_2$  or  $\text{NH}_3$

E500ii: Sodium bicarbonate (baking soda)

E503i: Ammonium bicarbonate  
(baker's ammonia)

E503ii: Ammonium carbonate  
(neutral ammonium carbonate)



# 15. Raising (or leavening) agents in baking

## B) Mixtures of raising agents (baking powders)

Powdered sodium bicarbonate with inert ingredients (starch, flour, calcium lactate, calcium carbonate) and another ingredient such as:

**E344 Citric acid**

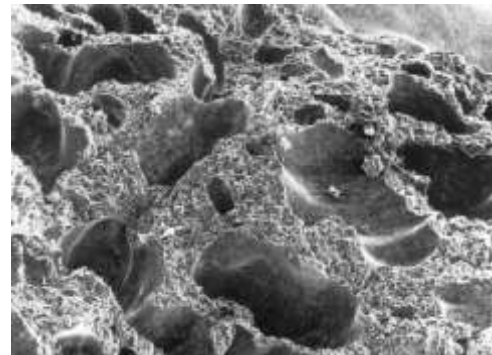
**E 336i Potassium bitartrate (cream of tartar)**

**Phosphoric salts (E339, E340, E341, E343, E450, E451, E452)**



## 15. Raising (or leavening) agents in baking

**Mode of action:** (a) Gas production after dissolution and in the presence of an acid (e.g. acid from milk, yogurt, orange juice) and/or (b) Gas production during baking.



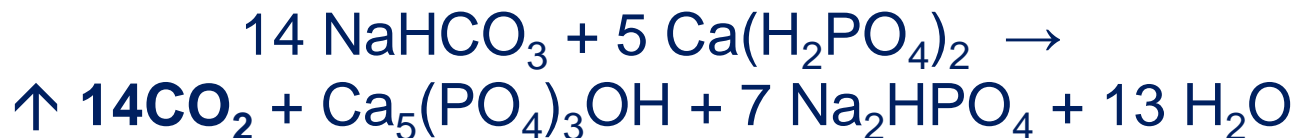
Gas (CO<sub>2</sub>)  
bubbles in  
the dough

## 15. Raising (or leavening) agents in baking

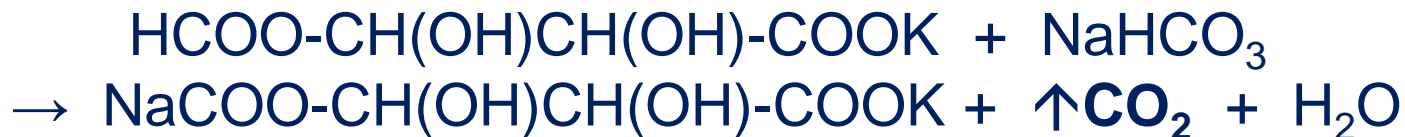
**Sodium bicarbonate** (baking soda) + **Acid** (vinegar, lemon juice, etc.):



**Sodium bicarbonate** (baking soda) + **Calcium dihydrogen phosphate**:



**Potassium bitartrate** (cream of tartar) + **Sodium bicarbonate** (baking soda):



## 16. Flavor enhancers

“Substances that enhance the existing taste and/or aroma of food”

E620 Glutamic acid

E621 Monosodium glutamate (MSG)

E622 Monopotassium glutamate

E623 Calcium glutamate

E624 Ammonium glutamate

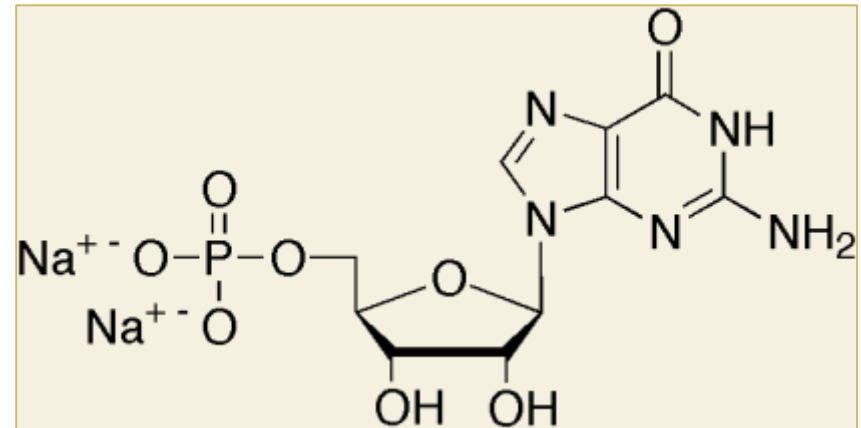
E625 Magnesium glutamate

E626 Guanylic acid

E627 Sodium guanylate

E628 Potassium guanylate

E629 Calcium guanylate





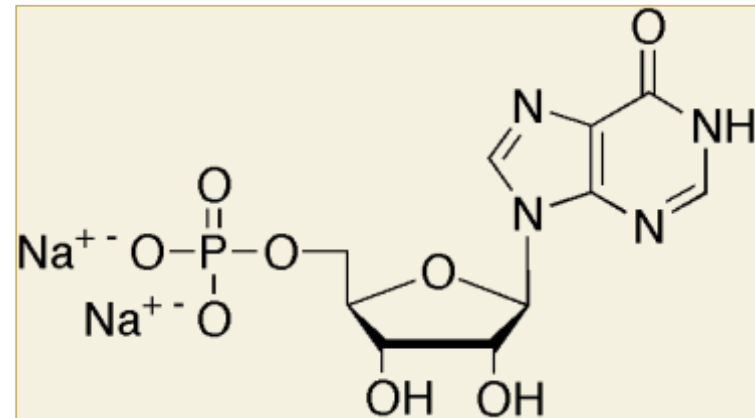
## 16. Flavor enhancers

**E630 Inosinic acid**

**E631 Sodium inosinate** →

**E632 Potassium inosinate**

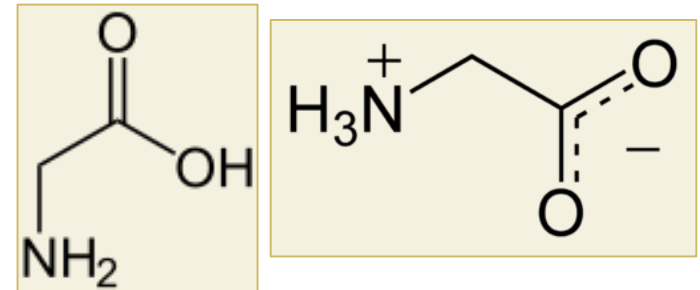
**E633 Calcium inosinate**



**E634,5 Salts of 5'-ribonucleotides with Ca and Na  
(mixtures of guanylates and inosinates)**

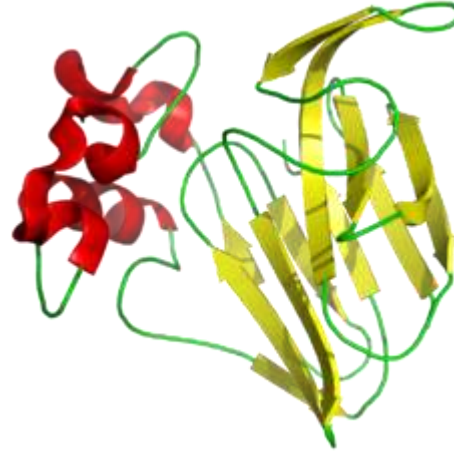
**E640 Glycine and its sodium salt** →

**E650 Zinc acetate**

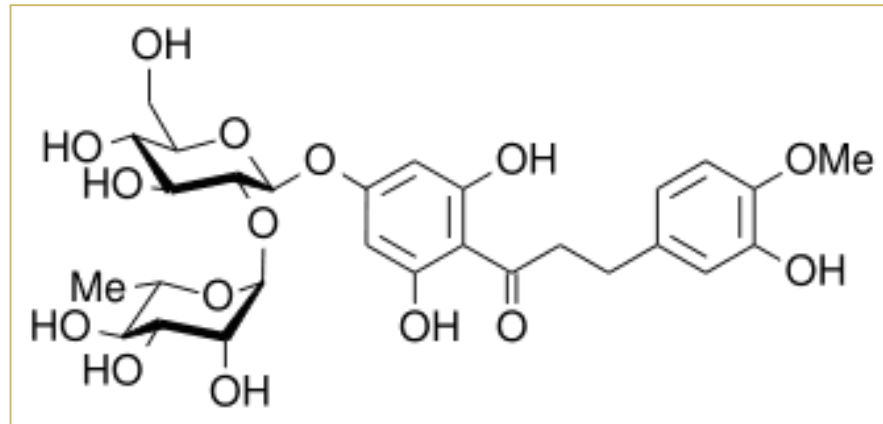


## 16. Flavor enhancers

**E957 Thaumatin**

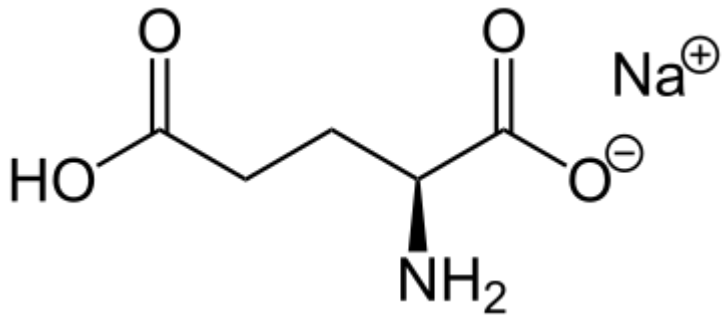


**E959 Neohesperidin (dihydrochalcone)**



## 16. Flavor enhancers

### E621 Monosodium glutamate (MSG)



- It is traditionally made from wheat gluten
- Today, produced by fermentation of starch, sugar, molasses, etc. with *Brevibacterium*, *Arthrobacter*, *Microbacterium* & *Corynebacterium* bacteria
- In Asian cuisine: seaweed extract, rich in glutamic acid is used

# 16. Flavor enhancers

## E621 Monosodium glutamate (MSG)



## 16. Flavor enhancers

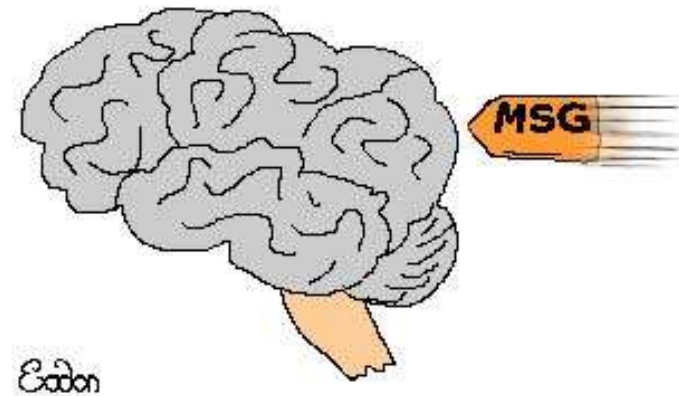
### E621 Monosodium glutamate (MSG)

#### Chinese Restaurant Syndrome (?)

(expressed in 1968 by Robert Ho Man Kwok: «Chinese restaurant syndrome», *New England Journal of Medicine*)

#### Symptoms:

- ✓ Onset: 15 to 20 min after consumption
- ✓ Duration: 2 hours
- ✓ Tingling sensation in the back of the throat and then in the hands and chest
- ✓ Weakness
- ✓ Rapid heartbeat





## 17. Chelating agents

“substances that form chelates with metal ions”

- ✓ Binding metals such as Cu, Ni, Fe
- ✓ Preventing degradation of food quality through the catalysis of oxidative reactions by metals

E385 Calcium disodium EDTA (Ethylenediaminetetraacetic acid)

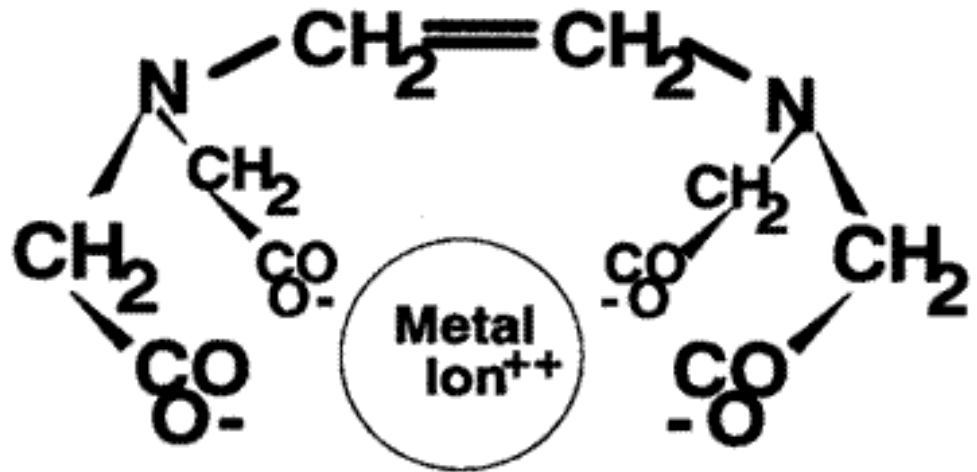
E509 Calcium chloride

E575 Glucono-delta-lactone

E576 Sodium gluconate

E577 Potassium gluconate

E333 Calcium citrates





*Thank you !*